

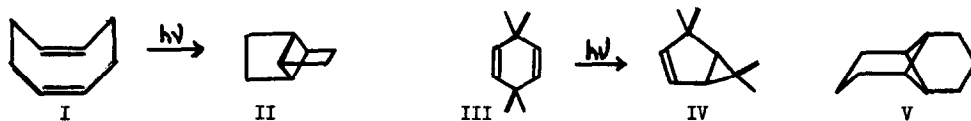
IRRADIATION OF THE cis,cis-CYCLODECA-1,6-DIENE SYSTEM

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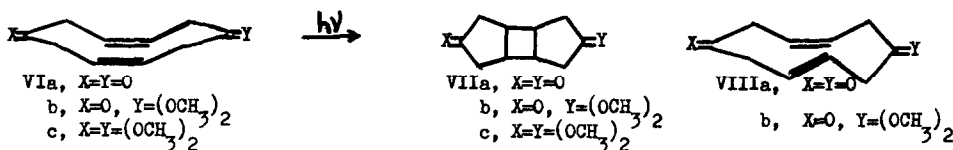
(Received in UK 14 June 1968; accepted for publication 16 September 1968)

The photochemical reaction of cis,cis-cycloocta-1,5-diene (I) yielded the cross-bonded tricyclo [3.3.0.0<sup>2,6</sup>] octane (II) (1) and that of the cyclohexa-1,4-diene III led to the rearranged bicyclic product IV (2). We wish to report the photochemical reaction of the cis,cis-cyclodeca-1,6-diene system (3), (4). This follows the Woodward-Hoffmann rule (5) for the photochemical



cycloaddition of two double bonds to yield cyclobutane VII, but not to the cross-bonded tricyclo [4.4.0.0<sup>2,7</sup>] decane system V (6), (7).

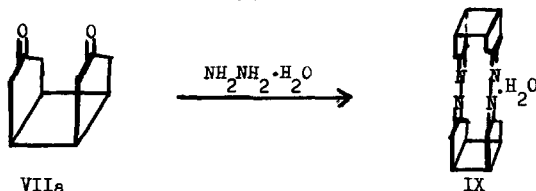
Irradiation (8) of 1 gr. of VIa in 130 ml benzene was carried out at room temperature under N<sub>2</sub> atmosphere. Thin layer chromatography indicated the completion of the reaction within 2 hours.



Two major products, isomeric with the starting material, were isolated by column chromatography on Florisil. The less polar compound isolated (mp 73-4°, 25-30%) was shown to be cis,trans-cyclodeca-3,8-diene-1,6-dione (VIIIa). [ $\nu_{\text{max}}^{\text{KBr}}$  1699, 982 and 706 cm<sup>-1</sup>; nmr (CDCl<sub>3</sub>)  $\delta$  5.90 (2H, multiplet, trans olefinic protons), 5.60 (2H, triplet, cis olefinic protons) and 3.60 - 3.12 (8H, multiplet, allylic and  $\alpha$ -carbonyl protons)]. Further confirmation of this structure was made by the catalytic reduction, using Adam's catalyst, to cyclodeca-1,6-dione, mp 98-9° (9). The more polar product (mp 133-4°, 45-50%) was shown to be the dione VIIa [ $\nu_{\text{max}}^{\text{KBr}}$  1725 cm<sup>-1</sup>,  $\nu_{\text{max}}^{\text{CHCl}_3}$  1727 cm<sup>-1</sup>; nmr  $\delta$  2.75 (4H, singlet, cyclobutane protons) and 2.60-2.45 (8H, multiplet,

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$\alpha$ -carbonyl protons, disappear upon exchange with deuterium). The cis, syn, cis structure of VIIa (10) was also chemically proved by reacting with hydrazine hydrate in 1:1 ratio to give IX (11). [mp  $> 290^\circ$  (dec.);  $\nu_{\text{max}}^{\text{KBr}}$  1655  $\text{cm}^{-1}$  (C=N)].



The monoketone VIb also reacts in the same manner as VIa giving VIIIb (mp  $72-3^\circ$ , 30%),  $\nu_{\text{max}}^{\text{KBr}}$  1719, 980 and 697  $\text{cm}^{-1}$ , nmr  $\delta$  5.50 (4H, multiplet), 3.27 (6H, singlet), 3.65-2.35 (8H) and VIIb [bp (bath temperature)  $70^\circ/0.5$  mm, 40%],  $\nu_{\text{max}}^{\text{neat}}$  1730, 1111 and 1047  $\text{cm}^{-1}$ , nmr  $\delta$  3.25 (3H, singlet), 3.18 (3H, singlet), 2.96-1.70 (12H).

Since VIc has no absorption in the UV (12), (13), acetone was used as sensitizer to give VIIC (mp  $56-7^\circ$ , 65%),  $\nu_{\text{max}}^{\text{KBr}}$  1089, 1049 and 869  $\text{cm}^{-1}$ , nmr  $\delta$  3.22 (6H, singlet), 3.15 (6H, singlet), 2.38 (4H, broad singlet, cyclobutane protons), 2.10-1.64 (8H).

Both VIIb and VIIC are hydrolyzed, in quantitative yield, to VIIa.

ACKNOWLEDGEMENT. - The author is grateful to Prof. R.Mechoulam for his encouragement and valuable discussions.

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3. The cis, cis-cyclodeca-1,6-diene system VI, in the boat conformation, is known to be prone to transannular interactions (4).
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5. R.Hoffmann and R.B.Woodward, *J. Am. Chem. Soc.*, **87**, 2046 (1965).
6. Structure V represents the basic carbon skeleton of some natural products such as copaene (7).
7. C.H.Heathcock, R.A.Badger and J.W.Patterson, Jr., *J. Am. Chem. Soc.*, **89**, 4133 (1967).
8. Hanau Q81 high pressure lamp, in a Pyrex deeper-well, was used.
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10. Stereochemical considerations of the cycloaddition are being investigated.
11. Satisfactory elemental analysis was obtained, as well as correct mass-spectrum with parent peak at 320 for the dimer (The  $\text{H}_2\text{O}$  molecule could not be detected).
12. There is a maximum at about 210  $\text{m}\mu$  ( $\epsilon \sim 2000$ ).
13. Irradiation in benzene for 9 hours gives unchanged starting material.